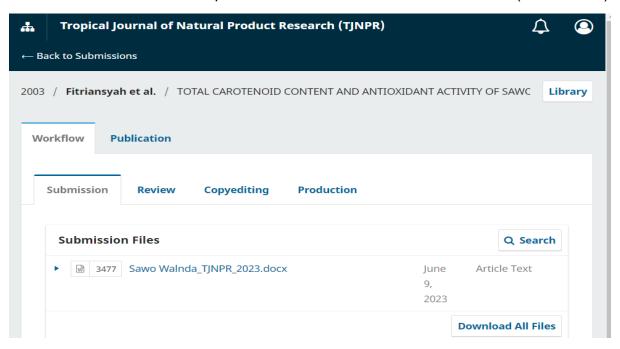
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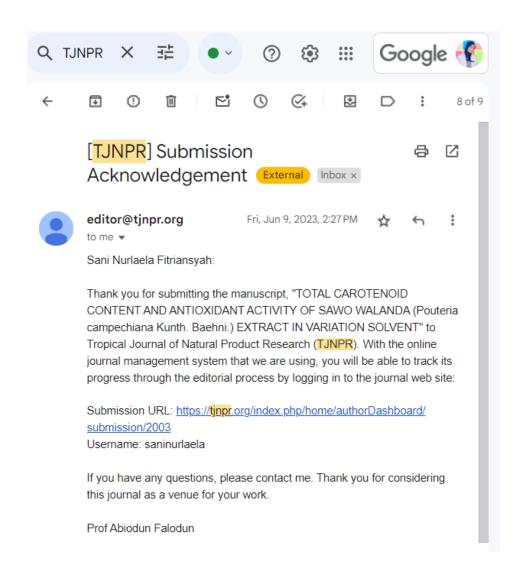
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Total Carotenoid Content and Antioxidant Activity of Sawo Walanda (Pouteria campechiana Kunth. Baehni) Extract from Various Solvent

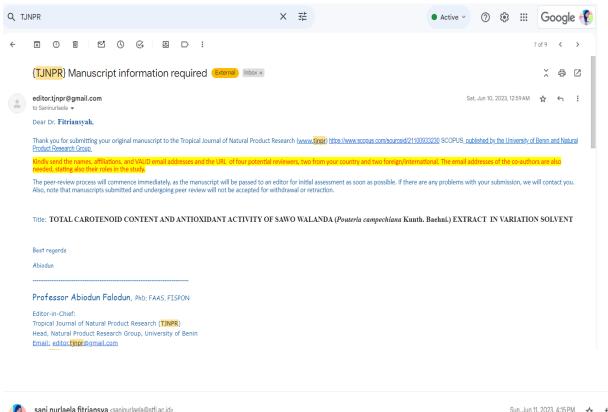
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- 3. Revisi pertama, Accepted dengan revisi sedang (moderate revision) (17 Juni 2023)
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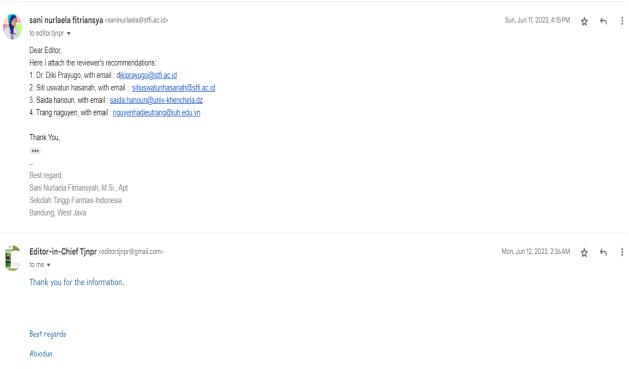
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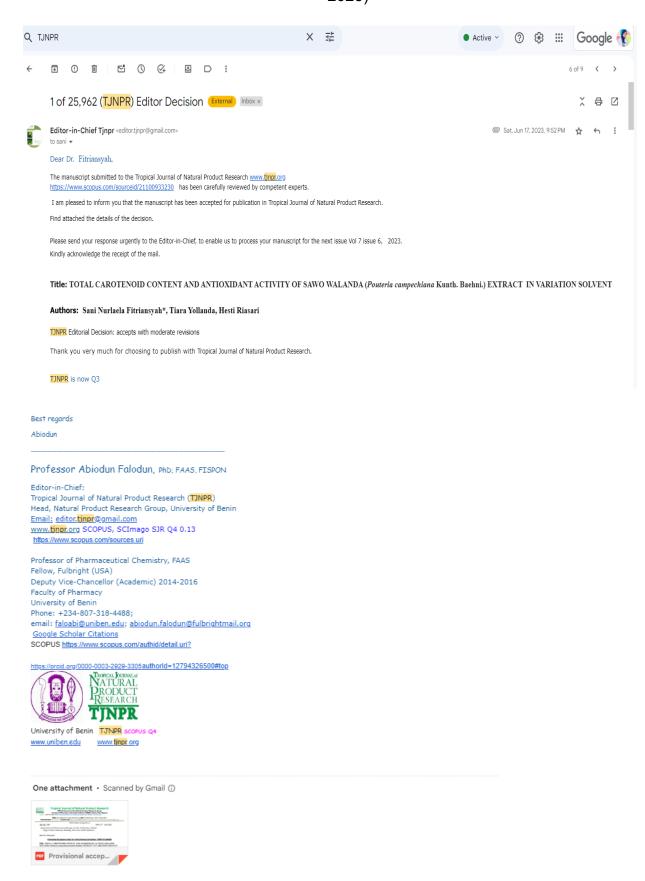


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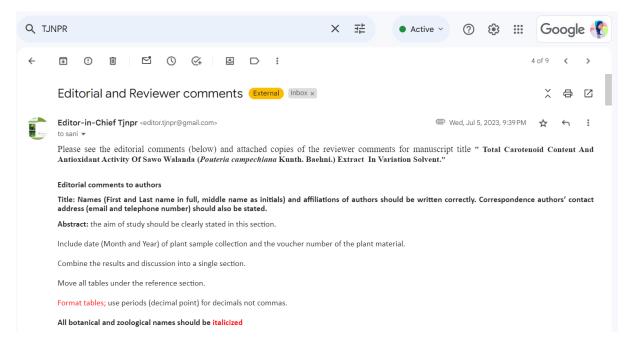




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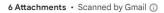


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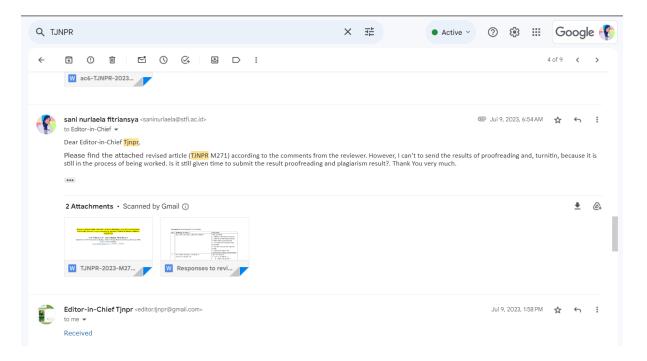








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TOTAL CAROTENOID CONTENT AND ANTIOXIDANT ACTIVITY OF SAWO WALANDA (*Pouteria campechiana* Kunth. Baehni) EXTRACT FROM VARIOUS SOLVENTS

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ABSTRACT

Sawo Pouteria campechiana, also known as sawo, walanda is rich in carotenoid. According to reports, carotenoid can act as antioxidants. Solvent selection is important in extracting the carotenoid.Indonesia, is known for its high abundance of carotenoids, which are recognized for their antioxidant properties., The objective of extraction of these constituents involves a crucial step of selecting an appropriate solvent, as indicated by several studies. Therefore, this study wasgims, to evaluateassess, the antioxidant activity and determined as well as determine the total carotenoid content of P. campechiana fruit and leaves leaf extract in various solvents, The research methods included the study methodology involved extraction used using the maceration method with n-hexane and ethyl acetate solvents, Antioxidant activity and determination of total carotenoid were carried out using UV-Vissible spectrophotometry. DPPH iswas used as a free radical and, while \(\beta\)-carotene is usedserved, as the standard for total carotenoid content. The findings revealed results showed, that the p-hexane fruit (NF) extract had the highest total carotenoid concentration (70.028 gBEQ/100g100 g, extract) and the), while n-hexane leaf (NL) extract had the lowest (39.540 gBEQ/100g loo g extract). Antioxidant activity was expressed with IC₅₀ (µg/ml). The result of antioxidant activity showed that the ethylEthyl, acetate leaf (EL) extract hadexhibited better antioxidant activity with IC_{50} (of 3.094 μ g/mL \pm 0.82) than the n hexane leaves compared to NL (9.270μg270 μg/mL±1.201), the ethyl acetate fruit extract(EF) (31.516μg516 μg/mL±1.786), and the n hexane fruit extractNF, (45.382µg382 µg/mL±2.31)-) extracts. Based on the statistical analysis results, the coefficient correlation of total carotenoid content with IC_{50} DPPH iswas r = 0.855, p < 0.01. This meansindicated that an increase in total carotenoid content isdid not always followed by an increaselead to a proportional increment in the inhibition of DPPH. In our work, it can be concluded These findings revealed that n-hexane solvent iswas, more suitable for extracting carotenoids from the fruit part, while ethyl acetate iswas more suitableappropriate for the leaves. Fruitleaf. Furthermore, the fruit, and leaf extract of P. campechiana havehad, potential antioxidant activity as natural ingredients in the food and pharmaceutical industries.

Key word Keywords: Sawo walanda, Pouteria campechiana, carotenoid, antioxidant activity

INTRODUCTION

The family Sapotaceae, which is extensively distributed globallyworldwide, includes the genus Pouteria. In traditional medicine, Pouteria species have been a rich history in traditional medicine, being used to treatfor the treatment of various ailments such as back pain, ulcers, skin eruptions, and inflammation. One of Among the Sapotaceae family's medicinal plants is within the Sapotaceae family, Pouteria campechiana—which was known, commonly referred to as sawo walanda in Indonesia, has been reported to hold significant importance.

According to a previous study, the fruit of *P. campechiana* (Kunth.). Baehni.) is a significant source of carotenoids² and is well knownhas gained recognition for its anti-inflammatory, antioxidant, and hepatoprotective properties.^{3,4} *Pouteria campechiana*—Furthermore, extracts obtained from its leaf and fruits extract contains phenolic group, flavonoid, and terpenoid group and—fruit have been found to contain phenolics, flavonoids, and terpenoids, exhibiting significant antioxidant activity.⁵

Sources Medicinal plants serve as a valuable source of raw materials with phenolic containing phenolics, flavonoids, other polyphenols, and terpenoids include medicinal plants. A phytochemical molecule may. These compounds have been reported to be the reason for a plant's biological action responsible for the various bioactivities exhibited by plants. The specific phytochemical component produced will depended on the extraction technique and the solvent employed. Carotenoids belongare compounds belonging to the group of secondary metabolites called known as terpenoids. The following Some of the common solvents

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are employed to extractused for the extraction of carotenoids: include acetone, chloroform, hexane, methanol, methylene chloride, ethyl acetate, and diethyl ether. The appropriateness of the chosenchoice of solvent will dependis largely dependent on the kindstype and concentrations of carotenoids that are intended for extraction. So choosing the rightbeing extracted. Therefore, the selection of the appropriate solvent for the extraction process becomes is very important in carotenoid extraction.

In. Therefore, this study, the aims to assess the antioxidant activity as well as determine the total carotenoid content of P. campechiana fruit and leaf extract in n hexane and ethyl acetate extract fruit and leaves P.campechiana was calculated, antioxidant activity were tested and the various solvents. The correlation of total carotenoid content with antioxidant activity were was analyzed use Pearson's method.

MATERIAL AND METHODS

Material

The materials used in this study included DPPH (2,2-diphenyl-1 pichydrazyl) (Sigma-Aldrich), ascorbic acid (Sigma-Aldrich), beta earotencarotene (β-carotene) (Sigma-aldrich), Sawo walanda (*Pouteria campechiana* Kunth. Baehni-). methanol, ethyl acetate, n-hexane, and other reagent were of analytical grades and grade reagents, which were obtained from Merck.

Sample preparation Preparation

<u>PouteriaP.</u> campechiana fruit and <u>leaves[eaf]</u> were freshly collected from Bandung, West Java, Indonesia, in January 2019. The sample was <u>determined identified</u> in the Herbarium Jatinangor, <u>Thethe</u> Plant Taxonomy Laboratorium, Biology Department, UNPAD, with <u>reference</u> number 123/HB/01/2020. <u>FruitSubsequently, the fruit</u> and <u>leaves are leaf were</u> sorted, washed, dried at 40°C – 45°C, and <u>grindedground</u> into powder.

Extraction

Three hundred A total of 300 grams of the powderpowdered fruit and leavesleaf was extracted byusing maceration. Each sample was extracted using n-hexane and ethyl acetate solvent, and the procedure was repeated in triplicate. EachFurthermore, each extract was concentrated byusing a rotary evaporator at 50°C and resulted to produce n-hexane extract fruit (NF), ethyl acetate fruit (EF), n-hexane extract leavesleaf (NL), and ethyl acetate extract leavesleaf (EL).) extracts.

Phytochemical analysis Analysis

<u>In order to To</u> assess the presence of secondary metabolites in *P. campechiana* fruit and <u>leaves[eaf]</u> extract, phytochemical screening was <u>conducted carried out</u> to <u>test determine</u> the presence of alkaloids, flavonoids, tannins, polyphenols, monoterpenes/sesquiterpenes, triterpenoids/steroid, quinones, and saponins.⁶

Antioxidant activity Activity

Antioxidant activity test was carried out using the DPPH adopted frommethod proposed by Fidrianny, (2018-). The standard for contrasting antioxidant chemicals was ascorbic acid. The and the concentration of DPPH used was 50 µg/mL, which servesserved as the control. The percentage of free radical inhibition was measured by mixing the DPPH solution with the sample in a (1:1) ratio and after incubation during. After incubating the sample for 30 min subsequently measuring, the absorbance was measured at 517 nm. This, and this procedure was repeated in triplicate for each sample.

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Determination of IC₅₀ Extract to DPPH

A calibration curve connecting the various sample concentrations to the percentage of sample inhibition against DPPH was used to calculate the IC₅₀ value against DPPH. <u>If The IC₅₀</u> was then determined using the calibration linear regression value. <u>The IC₅₀ value was calculated using</u>, with the x-value <u>from the linear regression</u> being used for calculations.

Determination of Antioxidant Activity Index

The antioxidant group of a sample <u>iswas</u> identified using the antioxidant activity index (AAI).⁸ (Scherer) The), which was calculated using the formula used to determine the AAI value is below:

$$AAI = rac{Final\ concentration\ of\ DPPH}{Final\ concentration\ of\ DPPH}$$
 $AAI = rac{Final\ concentration\ of\ DPPH}{Final\ concentration\ IC50}$

Determination of total carotenoid content Total Carotenoid Content

Thaipong and Fidrianny decided to measure proposed a measuring method to assess the overall carotenoid content. The extracts were diluted in n-hexane, and their absorbance was measured at a wavelength of 470 nm. The Furthermore, the analysis was conducted carried out in triplicate for each extract. To establish a standard curve, beta carotene was employed as a standard at concentrations ranging from 10 to 40 $\mu g/mL$. the The total carotenoid content was determined utilizing using the linear regression equation of the calibration curve, and the results were expressed as beta carotene equivalent of 100 g extract (g BE/100 g extract).

Data Analysis

Analyses of each sample were carried out in three copies. The triplicates, where the averages and standard deviations of at least three separate experiments arewere used to calculate all of the results. The Subsequently, Pearson's method was used to determine a correlation between the total carotenoid concentration and antioxidant activity. 10

RESULT AND DISCUSSION

Extraction

Extraction was carried out byusing maceration with different polarity solvents. The amount of metabolite content of each *P. campechiana* extract can be seen presented in Table 1. The findings showresults showed that ethyl acetate solvent maycould extract a higher amount of metabolites from *P. campechiana* fruit and leaves. Bothleaf. Furthermore, primary and secondary metabolites maycould be among the extracted compounds.

Phytochemical analysis Analysis

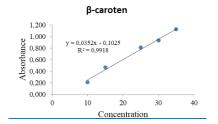
Table 2 <u>listsshows</u> the results of the phytochemical screening on the crude and each extract. Phenolic compounds, flavonoids, and saponin were absent in NF and NL_{7} , <u>while</u> Monoterpene/sesquiterpene was <u>absentant found</u> in EF and EL. Phenol and saponins <u>arewere</u> semipolar to polar metabolites, <u>whilebut</u> monoterpenes/sesquiterpenes <u>tendtended</u> to be

nonpolar to semipolar. <u>Several studies showed that P. campechiana is knownwas be</u> rich in phenolic compounds, flavonoids, and terpenoids. <u>P. campechiana The</u> leaf is rich in flavonoid, <u>stilbenoid, contained high levels of flavonoids stilbenoids</u> and <u>tannin¹¹, whereas intannins¹¹, while the fruit was rich in terpenoids, including carotenoid¹² and phenolic acid¹³. According to Sangeetha¹⁴-related to our work, which flavonoid, flavonoids and <u>saponinsaponins</u> were <u>also</u> absent in <u>the</u> fruit extract. Acetone extract leaf <u>P. campechiana presents alkaloid, flavonoid, saponin, steroid, tannincontained alkaloids, flavonoids, saponins, steroids, tannins, and terpenoid¹⁴terpenoids¹⁴.</u></u>

Carotenoid Total Content in Extract P._campechiana

Total carotenoid content was calculated based on the standard curve beta carotene with linear regression y=0.0352x-0.103, $R^2=0.992$ (Fig. 1). In measuring the total carotenoid content, beta-carotene iswas frequently employed as a standard. The highest total carotenoid-content was found in NF of $(70,028\pm2,341)$ g BE/100 g extract and was found in NF, while the smallest was indicated exhibited by NL at $(39,40\pm1,76)$ g BE/100 g extract, as shown in Fig. 2. The intensity of the carotenoids iswas attributed to their conjugated double bonds, which result inled to a vibrant yellow-orange shade. NF and EF gave stronger orange-yellow color than compared to NL and EL.

Carotenoids are referred to a class of compounds with both polar and nonpolar characteristics. The hydrocarbon carotenoids—(, namely beta carotene and lycopene) are were examples of the nonpolar category. Meanwhile, polar carotenoids—eategory. Polar carotenoids—(, such as lutein, canthaxanthin, astaxanthin, and fucoxanthin) when the were oxygenated derivativederivatives of hydrocarbon carotenoids—are, as shown in FigureFig. 3. ¹⁶ Carotenoids are were typically extracted using organic solvents, such acetone, chloroform, hexane, methanol, and diethyl ether. The sample matrix and its components, moisture content, and—the functional group (polarity)), and the chain length of the existing earotenoids are all compounds were significant factors induring the extraction of carotenoids... ¹⁶ The most frequent methods used for extracting both polar and nonpolar carotenoids simultaneously include included acetone, ethanol, ethyl acetate, and n-hexane... ¹⁷ Hexanewith hexane and acetone are being the common solvents most frequently for extracting carotenoids. ¹⁷



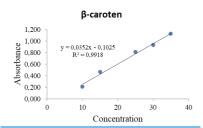
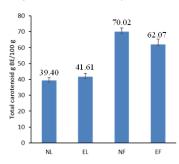


Fig. 1 Standard Curve β-carotene



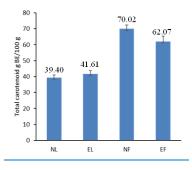


Fig. 2 Total carotenoid content PouteriaP. campechiana extract

P. The results showed that P. campechiana eonducted terpenoid in n-hexane and ethyl acetate fruit and leaf extracts contained terpenoids based on the phytochemical analysis. Category Carotenoids were a category of terpenoid found in the samples. A previous study showed that contains carotenoids. P. campechiana exhibited contained beta-carotene, beta cryptoxanthin, and violaxanthin, according to a prior study. (Fig. 43)¹⁸, and Furthermore, the total carotenoid content in petroleum ether fruit P. campechiana showed was 278 µg/g extract. 19

Fig. 3 The carotenoid in *P._campechiana*¹⁸

Antioxidant Activity and Correlation With Total Carotenoid Content

Antioxidant The antioxidant activity was expressed in terms of IC₅₀ and in Antioxidant Activity Index (AAI), as shown in Table 3. The IC₅₀ iswas the number of samples required to neutralize 50% of the free radical DPPH. The IC₅₀ value with, and the lowest value indicates indicated the strongest antioxidant activity. The IC₅₀ of DPPH scavenging capacities of NF, EF, NL, and EL were lower when compared to IC₅₀ of DPPH of ascorbic acid standard. Furthermore, LE had better antioxidant activity compared to NE, NF, and EF. The IC₅₀ DPPH of ascorbic acid was 1,98 μg/ml, the IC₅₀ DPPH values of NL, NL and EL ranged from 3.094 to 9.27 μg/mL, while NF₇ and EF ranged from 31,516 to 45,382 μg/mL.

According to some of the previous studies, the *PouteriaP* campechiana hashad antioxidant properties. In different extraction methods using Soxhlet with increasing polarity of the solvent, the The antioxidant activity of n-hexane, ethyl acetate, and ethanol leaf extract of leaves of *P. campechiana* the plant had been reported employing different extraction methods using Soxhlet with increasing polarity of the solvent. The Tollow ethanol 70% extract of pulp, peel, and leaves leaf of *P. campechiana* showed antioxidant activityalso exhibited similar properties. The campechiana fruit extracts in methanol, acetone, ethanol, and water at various stages of ripeness (4, 8, 12, 16, and 24 weeks) were found to be effective antioxidants against DPPH. The Depth 15

The carotenoid group <u>eancould</u> also contribute to <u>the</u> antioxidant activity <u>exhibited by the test plant</u>. ²² Comparing the carotenoid and phenolic groups, the phenolic group <u>hashad</u> a higher potential for <u>antioxidant properties</u>. this <u>activity</u>. ²³ Beta carotene <u>hashad</u> antioxidant properties because it <u>eontainscontained</u> conjugated double bonds. Through a chemical reaction between carotenoids and free radicals, the conjugated double-bond structure <u>iswas</u> directly damaged. <u>Carotenoids containFurthermore</u>, <u>carotenoids contained</u> long conjugated chains, <u>which makes them very reactive</u> <u>leading to the high reactivity of these compounds</u>. ²⁴ The phenolic group <u>eancontributed</u> to the antioxidant activity. Donating hydrogen atoms to DPPH,

phenolic groups, and flavonoids <u>maycould</u> reduce the damage caused by free radicals.²⁵ Both <u>einnamicCinnamic</u> acid and benzoic acids significantly <u>increase increased the</u> antioxidant activity. Compared to benzoic acid, cinnamic acid <u>may have had</u> a greater impact on <u>antioxidant activity these properties</u>.²⁶ The <u>existence of a biological activity approach to the content of secondary metabolites</u>, can be one reason that can explain the difference in IC₅₀ values <u>inof</u> NF, EF, NL, and EL was caused by the different biological activities of their secondary metabolites.

Fruit and leaves leaf extract Pouteria of P. campechiana have the potential ashad antioxidant activity properties. The grouping antioxidant activity empotentials, which could be expressed with AAI value. Based on the AAI value, in Table 3, NF and EF arewere included in the moderate group antioxidant activity, while NL and EL are included were in the very strong antioxidant activity group.

The percentage inhibition value of DPPH iswas inversely correlated with the IC_{50} value of DPPH. The percentage inhibition value of DPPH increases with a decrease in the DPPH IC_{50} value. Based on statistics, the Pearson's correlation coefficient between the total carotenoid total content of $P_{campechiana}$ fruit and leaves leaf extract with IC_{50} scavenging of DPPH gave a value positive and significant value (r = 0.855, p<0.01). It means, This indicated that the total carotenoid content in the fruit and leaf extract has nothed no correlation with inhibition against DPPH potent.

Conclusion

The study result concludedIn conclusion, the results showed that the n-hexaneNF extract fruitof Pouteria campechiana was ahad higher carotenoid total content, followed by ethyl acetateEF, EL, and NL extracts. Furthermore, EL extract fruit, ethyl acetate leaves, and n-hexane extract leaves Pouteria campechiana. The ethyl acetate leaf extract hashad better inhibition of DPPH. Based on statistics, total carotenoid content in the fruit and leaves leaf extract of Pouteria campechiana has not significantly contributed no significant contribution to the inhibition of DPPH.

Conflict of Interest

The Authorauthor declares no conflict of interest.

Author's Authors' Declaration

The <u>authorauthors</u> hereby <u>declares</u> declare that the work presented in this article is original and that any liability for a claim relating to the content <u>of this article</u> will be borne by them.

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Table 1. Yield Extract P. campechiana

Extract	Yield of extract (%)
NF	2.82
EF	5.16
NL	2.4

EL 5.74

Table -2. Phytochemical Screening of Crude Drug and Extract P. campechiana

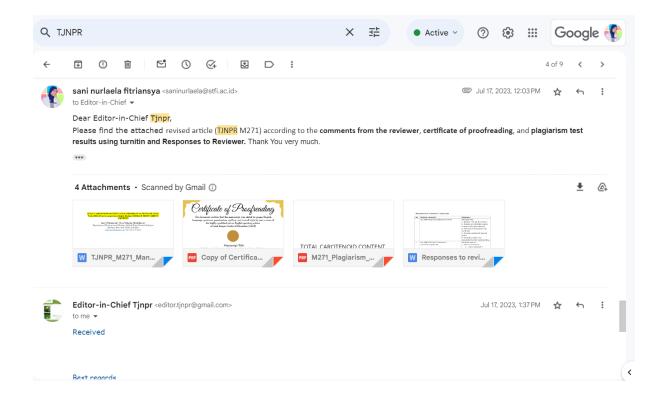
Metabolite	Crude Drug		Extract			
	Fruit	Leaf	NF	EF	NL	EL
Alkaloid	+	+	+	+	+	+
Phenol	+	+	-	+	-	+
Tanin	+	+	-	-	-	+
Flavonoid	+	+	-	+	-	+
Quinone	+	+	-	-	-	-
Saponin	-	-	-	-	-	-
Triterpenoid/steroid	+	+	+	+	+	+
Monoterpenoid/sesquiterpenoid	+	+	+	+	+	+

Table 3. Antioxidant activity with IC50 extract P. campechiana to DPPH

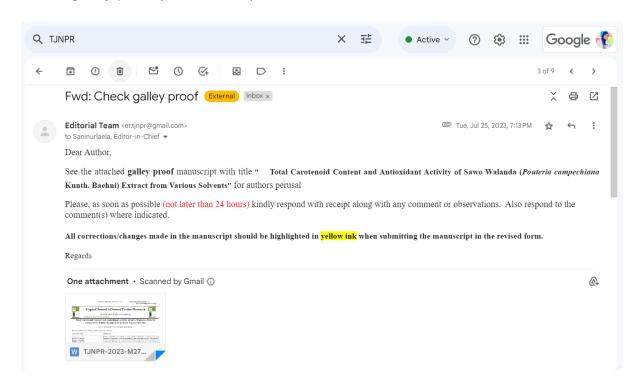
Sample	IC ₅₀ μg/mL	Antioxidant Activity Index
		(AAI)
NF	45.382 ± 2.31	0.551 ± 0.102
EF	31.516 ± 1.786	0.793 ± 0.098
NL	9.270 -± 1.201	2.697 ± 0.312
EL	$3.094 -\pm 0.82$	8.080 ± 0.221
As.acid	1.980 ± 0.230	12.626 ± 0.099

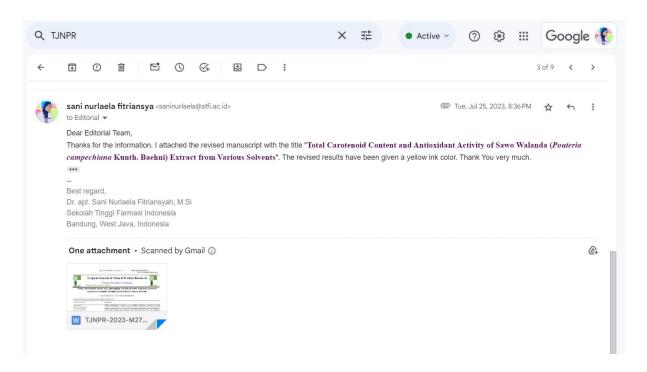
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Original Research Article

Total Carotenoid Content and Antioxidant Activity of Sawo Walanda (*Pouteria campechiana* Kunth. Baehni) Extract from Various Solvents

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ABSTRACT

Pouteria campechiana, also known as sawo walanda in Indonesia, is known for its high abundance of carotenoids, which are recognized for their antioxidant properties. The extraction of these constituents involves a crucial step of selecting an appropriate solvent, as indicated by several studies. Therefore, this study aims to assess the antioxidant activity as well as determine the total carotenoid content of P. campechiana fruit and leaf extract in various solvents. The study methodology involved extraction using the maceration method with n-hexane and ethyl acetate as solvents. Antioxidant activity and determination of total carotenoid were carried out using UV-Vissible spectrophotometry. DPPH was used as a free radical, while β-carotene served as the standard for total carotenoid content. The results showed that n-hexane fruit (NF) extract had the highest total carotenoid concentration (70.028 gBEQ/100 g extract), while n-hexane leaf (NL) extract had the lowest (39.540 gBEQ/100 g extract). Ethyl acetate leaf (EL) extract exhibited better antioxidant activity with IC₅₀ of 3.094 µg/mL±0.82 compared to NL (9.270 µg/mL±1.201), ethyl acetate fruit (EF) (31.516 μ g/mL \pm 1.786), and NF (45.382 μ g/mL \pm 2.31) extracts. Based on the statistical analysis results, the coefficient correlation of total carotenoid content with IC50 DPPH was r = 0.855, p<0.01. This indicated that an increase in total carotenoid content did not always lead to a proportional increment in the inhibition of DPPH. These findings revealed that n-hexane solvent was more suitable for extracting carotenoids from the fruit part, while ethyl acetate was more appropriate for the leaf. Furthermore, the fruit and leaf extract of P. campechiana had potential antioxidant activity as natural ingredients in the food and pharmaceutical industries.

Keywords: Sawo walanda, Pouteria campechiana, carotenoid, antioxidant activity

Introduction

The family Sapotaceae, which is distributed worldwide, includes the genus Pouteria. Pouteria species have a rich history in traditional medicine, being used for the treatment of various ailments such as back pain, ulcers, skin eruptions, and inflammation. Among the medicinal plants within the Sapotaceae family, *Pouteria campechiana*, commonly referred to as sawo walanda in Indonesia, has been reported to hold significant importance.

According to a previous study, the fruit of *P. campechiana* (Kunth. Baehni) is a significant source of carotenoids² and has gained recognition for its anti-inflammatory, antioxidant, and hepatoprotective properties.^{3,4} Furthermore, extracts obtained from its leaf and fruit have been found to contain phenolics, flavonoids, and terpenoids, exhibiting significant antioxidant activity.⁵

Medicinal plants serve as a valuable source of raw materials containing phenolics, flavonoids, polyphenols, and terpenoids. These compounds have been reported to be responsible for the various bioactivities exhibited by plants. The specific phytochemical component produced depends on the extraction technique and the solvent employed. Carotenoids are compounds belonging to a group of secondary metabolites known as terpenoids.

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Some of the common solvents used for the extraction of carotenoids include acetone, chloroform, hexane, methanol, methylene chloride, ethyl acetate, and diethyl ether.⁵ The choice of solvent is largely dependent on the type and concentrations of carotenoids being extracted. Therefore, the selection of the appropriate solvent is very important. Therefore, this study aims to assess the antioxidant activity as well as determine the total carotenoid content of P. campechiana fruit and leaf extract in various solvents. The correlation of total carotenoid content with antioxidant activity was analyzed using Pearson's method.

Material and Methods

Material

The materials used in this study included DPPH (2,2-diphenyl-1 pichydrazyl) (Sigma-Aldrich), ascorbic acid (Sigma-Aldrich), beta carotene (β -carotene) (Sigma-aldrich), Sawo walanda (*Pouteria campechiana* Kunth. Baehni), methanol, ethyl acetate, n-hexane, and other analytical grade reagents, which were obtained from Merck.

Sample preparation

P. campechiana fruit and leaf were freshly collected from Bandung, West Java, Indonesia, in January 2019. The sample was identified in the Herbarium Jatinangor, the Plant Taxonomy Laboratorium, Biology Department, UNPAD, with reference number 123/HB/01/2020. Subsequently, the fruit and leaf were sorted, washed, dried at 40°C – 45°C, and ground into powder.

Extraction

A total of 300 grams of the powdered fruit and leaf was extracted using maceration. Each sample was extracted using n-hexane and ethyl acetate solvent, and the procedure was repeated in triplicate. Furthermore, each extract was concentrated using a rotary evaporator at 50°C to produce n-hexane fruit (NF), ethyl acetate fruit (EF), n-hexane leaf (NL), and ethyl acetate leaf (EL) extracts.

Phytochemical analysis

To assess the presence of secondary metabolites in *P. campechiana* fruit and leaf extract, phytochemical screening was carried out to determine the presence of alkaloids, flavonoids, tannins, polyphenols, monoterpenes/sesquiterpenes, triterpenoids/steroid, quinones, and saponins.⁶

Antioxidant Activity

Antioxidant activity test was carried out using the DPPH method proposed by Fidrianny, (2018). The standard for contrasting antioxidant chemicals was ascorbic acid, and the concentration of DPPH used was $50\,\mu\text{g/mL}$, which served as the control. The percentage of free radical inhibition was measured by mixing the DPPH solution with the sample in a (1:1) ratio. After incubating the sample for 30 min, the absorbance was measured at 517 nm, and this procedure was repeated in triplicate for each sample.

% Inhibition =
$$\frac{(Blank \ abs - Sample \ abs)}{Blank \ abs} \ x \ 100\%$$

Determination of IC50 Extract to DPPH

A calibration curve connecting the various sample concentrations to the percentage of sample inhibition against DPPH was used to calculate the IC_{50} value against DPPH. The IC_{50} was then determined using the calibration linear regression value, with the x-value being used for calculations.

Determination of Antioxidant Activity Index

The antioxidant group of a sample was identified using the antioxidant activity index (AAI).⁸ (Scherer), which was calculated using the formula below:

$$AAI = \frac{Final\ concentration\ of\ DPPH}{Final\ concentration\ IC50}$$

Determination of Total Carotenoid Content

Thaipong⁹ and Fidrianny¹⁰ proposed a measuring method to assess the overall carotenoid content. The extracts were diluted in n-hexane, and their absorbance was measured at a wavelength of 470 nm. Furthermore, the analysis was carried out in triplicate for each extract. To establish a standard curve, beta carotene was employed as a standard at concentrations ranging from 10 to 40 μ g/mL. The total carotenoid content was determined using the linear regression equation of the calibration curve, and the results were expressed as beta carotene equivalent of 100 g extract (g BE/100 g extract).

Data analysis

Analyses of each sample were carried out in triplicates, where the averages and standard deviations of at least three separate experiments were used to calculate all the results. Subsequently, Pearson's method was used to determine a correlation between the total carotenoid concentration and antioxidant activity.¹⁰

Result and Discussion

Extraction

Extraction was carried out using maceration with different polarity solvents. The amount of metabolite content of each *P. campechiana* extract is presented in Table 1. The results showed that ethyl acetate solvent could extract a higher amount of metabolites from *P. campechiana* fruit and leaf. Furthermore, primary and secondary metabolites could be among the extracted compounds.

Phytochemical analysis

Table 2 shows the results of the phytochemical screening on the crude and each extract. Phenolic compounds, flavonoids, and saponin were absent in NF and NL, while Monoterpene/sesquiterpene was not found in EF and EL. Phenol and saponins were semipolar to polar metabolites, but monoterpenes/sesquiterpenes tended to be nonpolar to semipolar. Several studies showed that *P. campechiana* was be rich in phenolic compounds, flavonoids, and terpenoids. The leaf contained high levels

of flavonoids stilbenoids and tannins¹¹, while the fruit was rich in terpenoids, including carotenoid¹² and phenolic acid¹³. According to Sangeetha¹⁴, flavonoids and saponins were also absent in the fruit extract. Acetone extract leaf *P. campechiana* contained alkaloids, flavonoids, saponins, steroids, tannins, and terpenoids.¹⁴

Carotenoid Total Content in Extract P. campechiana

Total carotenoid content was calculated based on the standard curve beta carotene with linear regression y=0.0352x - 0.103, $R^2 = 0.992$ (Figure 1). In measuring the total carotenoid content, beta-carotene was frequently employed as a standard. The highest content of (70,028 ± 2,341) g BE/100 g extract was found in NF, while the smallest was exhibited by NL at $(39,40 \pm 1,76)$ g BE/100 g extract, as shown in (Figure 2). The intensity of the carotenoids was attributed to their conjugated double bonds, which led to a vibrant yellow-orange shade. 15 NF and EF gave stronger orange-yellow color compared to NL and EL. Carotenoids referred to a class of compounds with both polar and nonpolar characteristics. The hydrocarbon carotenoids, namely beta carotene and lycopene were examples of the nonpolar category. Meanwhile, polar carotenoids, such as lutein, canthaxanthin, astaxanthin, and fucoxanthin were oxygenated derivatives of hydrocarbon carotenoids, as shown in (Figure 3).16 Carotenoids were typically extracted using organic solvents, such acetone, chloroform, hexane, methanol, and diethyl ether. The sample matrix and its components, moisture content, the functional group (polarity), and the chain length of the existing compounds were significant factors during the extraction. 16 The most frequent methods used for extracting both polar and nonpolar carotenoids simultaneously included acetone, ethanol, ethyl acetate, and n-hexane, 17 with hexane and acetone being the common solvents.17

The results showed that *P. campechiana* n-hexane and ethyl acetate fruit and leaf extracts contained terpenoids based on the phytochemical analysis. Carotenoids were a category of terpenoids found in the samples. A previous study showed that *P. campechiana* contained betacarotene, beta-cryptoxanthin, and violaxanthin (Fig. 3)¹⁸. Furthermore, the total carotenoid content in petroleum ether fruit *P. campechiana* was $278 \mu g/g$ extract.¹⁹

Table 1: Yield Extract P. campechiana

Extract	Yield of Extract (%)
NF	2.82
EF	5.16
NL	2.4
EL	5.74

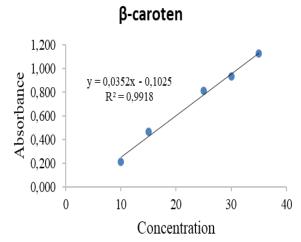


Figure 1: Standard Curve β-carotene

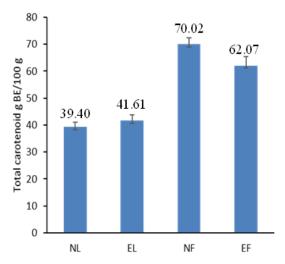


Figure 3: The carotenoid in P. campechiana¹⁸

Antioxidant Activity and Correlation With Total Carotenoid Content The antioxidant activity was expressed in terms of IC $_{50}$ and Antioxidant Activity Index (AAI), as shown in Table 3. The IC $_{50}$ was the number of samples required to neutralize 50% of the free radical DPPH, and the lowest value indicated the strongest antioxidant activity. The IC $_{50}$ of DPPH scavenging capacities of NF, EF, NL, and EL were lower compared to DPPH of ascorbic acid standard. Furthermore, LE had better antioxidant activity compared to NE, NF, and EF. The IC $_{50}$ DPPH of ascorbic acid was 1,98 μ g/ml, NL and EL ranged from 3.094 to 9.27 μ g/mL, while NF and EF ranged from 31,516 to 45,382 μ g/mL.

According to previous studies, *P. campechiana* had antioxidant properties. The antioxidant activity of n-hexane, ethyl acetate, and ethanol leaf extract of the plant had been reported employing different extraction methods using Soxhlet with increasing polarity of the solvent.²⁰ The 70% ethanol extract of pulp, peel, and leaf of *P. campechiana* also exhibited similar properties.²¹ *P. campechiana* fruit

extracts in methanol, acetone, ethanol, and water at various stages of ripeness (4, 8, 12, 16, and 24 weeks) were found to be effective antioxidants against DPPH.¹⁵

The carotenoid group could also contribute to the antioxidant activity exhibited by the test plant. ²² Comparing the carotenoid and phenolic groups, the phenolic group had a higher potential for this activity. ²³ Beta carotene had antioxidant properties because it contained conjugated double bonds. Through a chemical reaction between carotenoids and free radicals, the conjugated double-bond structure was directly damaged. Furthermore, carotenoids contained long conjugated chains, leading to the high reactivity of these compounds. ²⁴ The phenolic group also contributed to the antioxidant activity. Donating hydrogen atoms to DPPH, phenolic groups, and flavonoids could reduce the damage caused by free radicals. ²⁵ Cinnamic acid and benzoic acids significantly increased the antioxidant activity. Compared to benzoic acid, cinnamic acid had a greater impact on these properties. ²⁶ The difference in IC₅₀ values of NF, EF, NL, and EL was caused by the different biological activities of their secondary metabolites.

Fruit and leaf extract of *P. campechiana* had antioxidant activity potentials, which could be expressed with AAI value. Based on the AAI value in Table 3, NF and EF were included in the moderate group, while NL and EL were in the very strong group.

The percentage inhibition value of DPPH was inversely correlated with the IC₅₀ value of DPPH. Based on statistics, the Pearson's correlation coefficient between the total carotenoid total content of *P. campechiana* fruit and leaf extract with IC₅₀ scavenging of DPPH gave a positive and significant value (r = 0.855, p<0.01). This indicated that the total carotenoid content in the fruit and leaf extract had no correlation with inhibition against DPPH potent.

Conclusion

In conclusion, the results showed that NF extract of *Pouteria campechiana* had higher carotenoid total content, followed by EF, EL, and NL extracts. Furthermore, EL extract had better inhibition of DPPH. Based on statistics, total carotenoid content in the fruit and leaf extract of *Pouteria campechiana* had no significant contribution to the inhibition of DPPH.

 Table 2: Phytochemical Screening of Crude Drug and Extract P. campechiana

Metabolite	Crude Drug			act		
	Fruit	Leaf	NF	EF	NL	EL
Alkaloid	+	+	+	+	+	+
Phenol	+	+	-	+	-	+
Tanin	+	+	-	-	-	+
Flavonoid	+	+	-	+	-	+
Quinone	+	+	-	-	-	-
Saponin	-	-	-	-	-	-
Triterpenoid/steroid	+	+	+	+	+	+
Monoterpenoid/sesquiterpenoid	+	+	+	+	+	+

Table 3: Antioxidant activity with IC₅₀ extract *P. campechiana* to DPPH

Sample	IC ₅₀ μg/mL	Antioxidant Activity Index
		(AAI)
NF	45.382 ± 2.31	0.551 ± 0.102
EF	31.516 ± 1.786	0.793 ± 0.098
NL	9.270 ± 1.201	2.697 ± 0.312
EL	3.094 ± 0.82	8.080 ± 0.221
As.acid	1.980 ± 0.230	12.626 ± 0.099

Conflict of Interest

The authors declare no conflict of interest.

Authors' Declaration

The authors hereby declare that the work presented in this article is original and that any liability for claims relating to the content of this article will be borne by them.

Acknowledgments

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